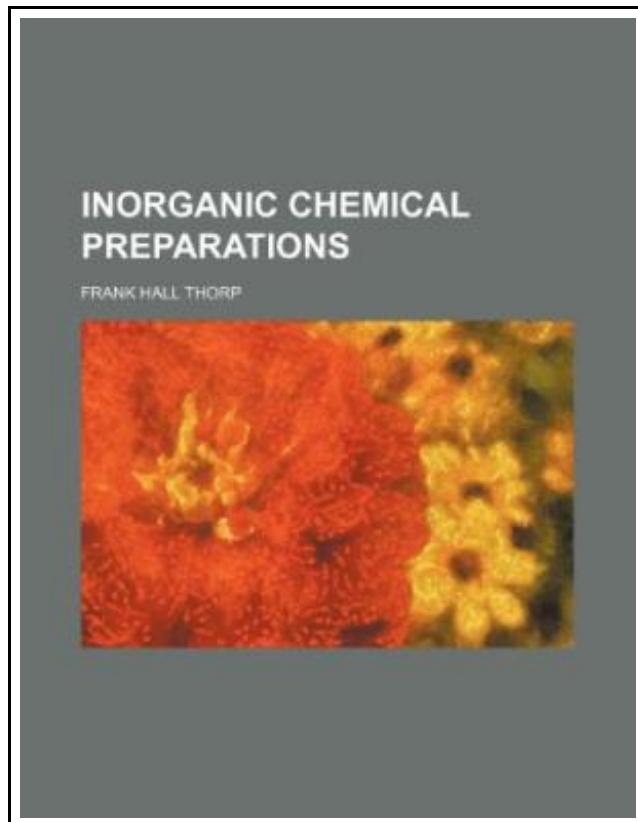


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Reviews

Merely no phrases to spell out. I actually have read through and i am certain that i will gonna study once again again later on. You wont truly feel monotony at at any time of your time (that's what catalogues are for about should you check with me).

(Jaiden Konopelski)

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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 54 pages. Dimensions: 9.7in. x 7.4in. x 0.1in. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1896 Excerpt: . . . t Poggiale, Ann. Chim. Phys. 3 8, 463. t Mulder, Scheik. Verhand. 1864, 79. The specific gravity of the taining 1 CUSO₄·5 H₂O 10 15 20 25 30 aqueous solution at 18 C. con H₂O is 1. 0063. 1. 0319. 1. 0649. 1. 0993. 1. 1354. 1. 1738. 1. 2146. Copper sulphate forms triclinic crystals of a dark blue color. When heated to 200 C. , all the water of crystallization is expelled and a white powder is left which is exceedingly hygroscopic. Calcined copper sulphate is often used to abstract water from alcohol and other liquids in which it is insoluble. Crystallized copper sulphate is used to some extent in dyeing, as a preservative agent, as the source of other copper salts, etc. 39. FERRIC CHLORIDE. FeCl₃·6 H₂O or Fe₂Cl₆·12 H₂O. M. W. 270. 5. (a)f Take of Iron (nails or bands), 75 gr. Hydrochloric acid, 36 Tw. (sp. gr. 1. 180), 345 cc. Nitric acid, 68 Tw. (sp. gr. 1. 340), 30 Water, 150 Put the iron into a liter flask with 227 cc. of the hydrochloric acid which has been diluted with 125 cc. of water. The solu Schiff, Gerlach, Zeit. anal. Chem. 8, 288. t After U. S. Pharmacopoeia, 1890. tion of the iron is slow at first but soon becomes very active, and the liquid becomes pale green in color. Cover the mouth of the flask with a watch glass, and let it stand on the steam table until all action...

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